



# Effects of annealing on near-infrared shielding properties of Cs-doped tungsten oxide thin films deposited by electron beam evaporation



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## ABSTRACT

Near-infrared (NIR) shielding properties are important for solar films. In this study, Cs<sub>x</sub>WO<sub>3</sub> films prepared using electron beam evaporation were characterized using X-ray diffraction, X-ray photoelectron spectroscopy, and spectrophotometry. The effects of annealing on NIR shielding properties and film microstructure were investigated. The results show that the NIR shielding properties of Cs<sub>x</sub>WO<sub>3</sub> films can be improved by annealing at 300–450 °C under pure H<sub>2</sub> atmosphere, the amorphous thin films being transformed to crystalline films. The Cs<sub>x</sub>WO<sub>3</sub> films annealed at 450 °C in pure H<sub>2</sub> atmosphere showed high transmittance of visible light (70%) and high NIR shielding ratio (99%).

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## 1. Introduction

Near-infrared (NIR) light shielding materials have been applied in solar collectors, smart windows, and optical filters [1–3]. NIR shielding has been achieved by utilizing transparent thermal coatings on windows. In summer, transparent thermal insulation coatings, which have excellent visible light transmittance, prevent heat transmission from the outside to the inside reducing air conditioning usage and, thus, energy consumption [4,5]. Transparent thermal insulation coatings can also be used as a solar filter. Transparent conductive oxides (TCOs) include indium tin oxide (ITO) and antimony tin oxide (ATO) [6]. Tungsten bronzes have received a lot of attention due to their electrochromic, photochromic, gasochromic, and superconducting properties [7–11]. Tungsten bronzes M<sub>x</sub>WO<sub>3</sub> with dopant ions such as Na, K, Rb, Cs, and other alkali metals [12,13] have been found to have the best optical and electrical properties. Takeda et al. [14] found that cesium tungsten bronze (Cs<sub>x</sub>WO<sub>3</sub>) film has excellent NIR absorption and shielding properties. One study [15] found that Cs<sub>x</sub>WO<sub>3</sub> has a strong NIR shielding ability as well as high transparency in the visible light region. Cs<sub>x</sub>WO<sub>3</sub> with a hexagonal structure exhibited higher NIR shielding than that of ITO, and is thus a possible replacement for ITO and ATO in windows. Cs<sub>x</sub>WO<sub>3</sub> powders have mainly been synthesized via the solvothermal reaction method [16–18] or hydrothermal reaction method [19–21] before being coated onto substrates. The present study is

aimed at the investigation of the H<sub>2</sub> annealing effects on the microstructure, morphology and NIR shielding properties of the Cs<sub>x</sub>WO<sub>3</sub> films prepared by electron beam evaporation.

## 2. Experimental Procedure

Thin films of Cs<sub>x</sub>WO<sub>3</sub> were deposited on quartz glass via electron beam evaporation. The source material was Cs<sub>x</sub>WO<sub>3</sub> (0 < x < 1) powders. (NanoStar Technology Co). The substrates were cleaned with acetone and methanol by treatment for 10 min in each solvent. Then, the substrates were washed with isopropyl alcohol and deionized water for 10 min. The electron gun was operated at 8 kV and 10 mA. The evaporation was carried out for 40 min at a base pressure of 7.5 × 10<sup>-6</sup> Torr. The films were grown at a constant temperature (273 K). The thickness of the Cs<sub>x</sub>WO<sub>3</sub> films was controlled to be 280 nm with deposition at a constant rate of 3 Å · s<sup>-1</sup> using an in situ quartz crystal thickness monitor. The phase composition and crystallinity of the Cs<sub>x</sub>WO<sub>3</sub> film were determined using X-ray diffraction (XRD) analysis (Rigaku DMAX 2500) with Cu Kα radiation (λ = 1.54 Å) at a scan speed of 1°/min. Ultrahigh-resolution field-emission scanning electron microscopy (UHRFE -SEM; ZEISS SUPRA series) and energy-dispersive X-ray spectroscopy (EDX) analysis were performed. The binding energies of tungsten core levels in the Cs<sub>x</sub>WO<sub>3</sub> film were determined using X-ray photoelectron spectroscopy (XPS; VersaProbe PHI 5000). The morphology of the film was determined using field-emission SEM (Hitachi-4800). The optical transmittance spectra were measured at room temperature (RT) using ultraviolet/visible/NIR spectrophotometry

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(Hitachi U-4100) in the wavelength range of 260–2600 nm. The surface morphology of the deposited film was observed using atomic force microscopy (AFM; Veeco CP-II) in semi-contact mode with Si tips at a scan rate of 1 Hz.

### 3. Results and Discussion

#### 3.1. Microstructure of films

Fig. 1 shows SEM images of  $\text{Cs}_x\text{WO}_3$  thin films before and after  $\text{H}_2$  annealing at 300–500 °C. No obvious precipitates or impurities appear on the surface of the  $\text{Cs}_{0.32}\text{WO}_3$  thin films (Fig. 1(a)). Fig. 1(b) and (c) show no distinct variation in the surface morphology of the thin films after annealing. However, when the annealing temperature was 400 °C, the grain size increased (Figs. 1(d) and (e)).

Fig. 2 shows three-dimensional (3D) AFM images (scanning area:  $3\ \mu\text{m} \times 3\ \mu\text{m}$ ) of  $\text{Cs}_x\text{WO}_3$  thin films deposited at various annealing temperatures. The films deposited at RT had a relatively smooth surface with vague grain boundaries, as shown in Fig. 2(a). The roughness was found to be 0.95 nm and the surface was continuous. The surface roughness and grain size increased with increasing annealing temperature. As shown in Figs. 2(b) and (c), when the annealing temperature was increased from 300 to 350 °C, the surface roughness increased from 1.55 to 1.73 nm. The surface roughness increased from 2.15 nm (400 °C) to 4.77 nm (450 °C). The grain size increase with annealing temperature is attributed to atomic rearrangement (atoms move faster on the surface of the films at higher temperature).

#### 3.2. Phase composition of films

Fig. 3 shows XRD patterns of  $\text{Cs}_x\text{WO}_3$  thin films before and after  $\text{H}_2$  annealing at 300–450 °C for 2 h. Fig. 3(a) shows XRD patterns of as-deposited  $\text{Cs}_{0.32}\text{WO}_3$  thin films. The diffraction peak absence indicates the amorphous film state. Figs. 3 (b–e) show XRD patterns of films annealed at 300, 350, 400, and 450 °C for 2 h, respectively. The  $\text{Cs}_x\text{WO}_3$  thin film crystallization is found after treatment at 300 °C. The characteristic peaks of the  $\text{Cs}_x\text{WO}_3$  thin films corresponded to the (002), (200), (112), (202), (212), (220), (204), (312), (400), and (224) planes. This agrees well with hexagonal  $\text{Cs}_{0.32}\text{WO}_3$  (JCPDS card no. 831334). No impurity peaks appeared. With increasing annealing temperature, the diffraction peaks became more apparent and sharper. It was reported that Cs ions (0.170 nm) can occupy hexagonal vacant tunnels (0.163 nm) in the hexagonal tungsten bronze structure [1]. The grain size,  $D$ , of the crystallites was calculated using Scherrer's formula:

$$D = \frac{K\lambda}{\beta \cos\theta}$$

where  $K$  is the Scherrer constant ( $=0.9$ ),  $\lambda$  is the wavelength of the X-ray ( $=1.54\ \text{\AA}$  for Cu  $K\alpha$  radiation),  $\theta$  is the Bragg angle, and  $\beta$  is the line broadening at half the maximum intensity (full width at half maximum). The grain size,  $D$ , increased from 15 to 30 nm when the annealing temperature was increased from 0 to 400 °C. The initially amorphous  $\text{Cs}_{0.32}\text{WO}_3$  thin films gradually crystallized with increasing  $\text{H}_2$  annealing temperature. This result is consistent with the SEM and AFM images in Figs. 1 and 2.

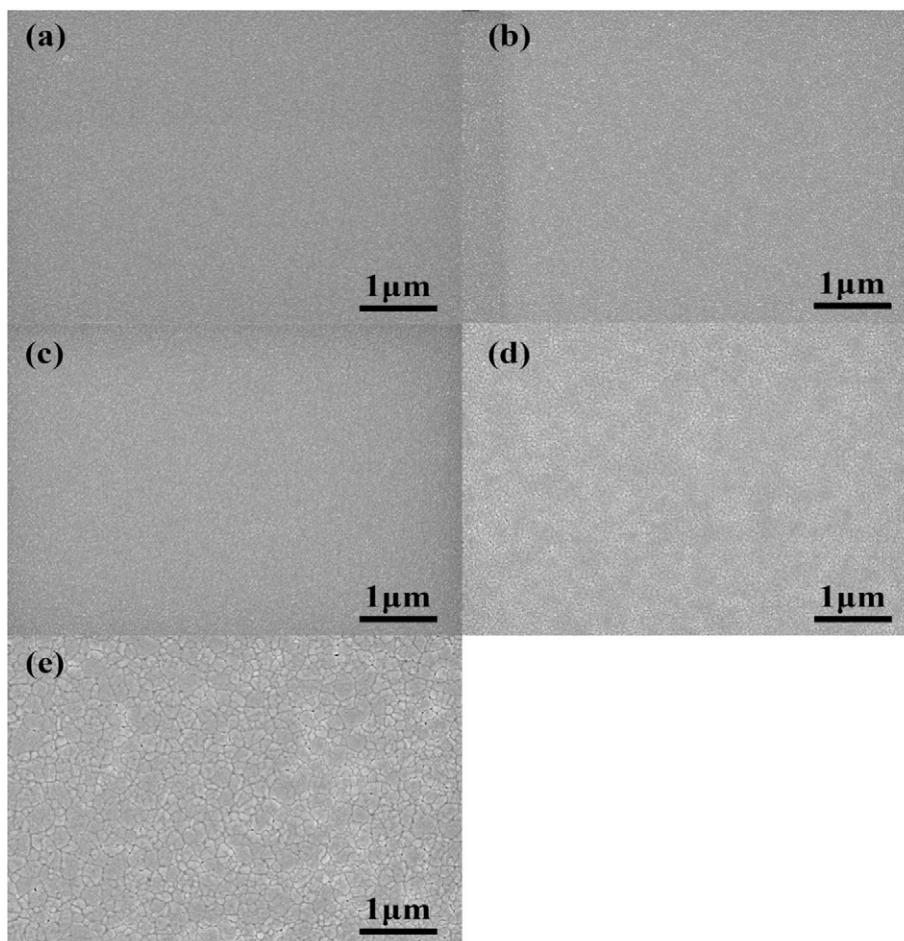


Fig. 1. SEM images of (a) as-sputtered  $\text{Cs}_{0.32}\text{WO}_3$  films and those annealed at (b) 300 °C, (c) 350 °C, (d) 400 °C, and (e) 450 °C.

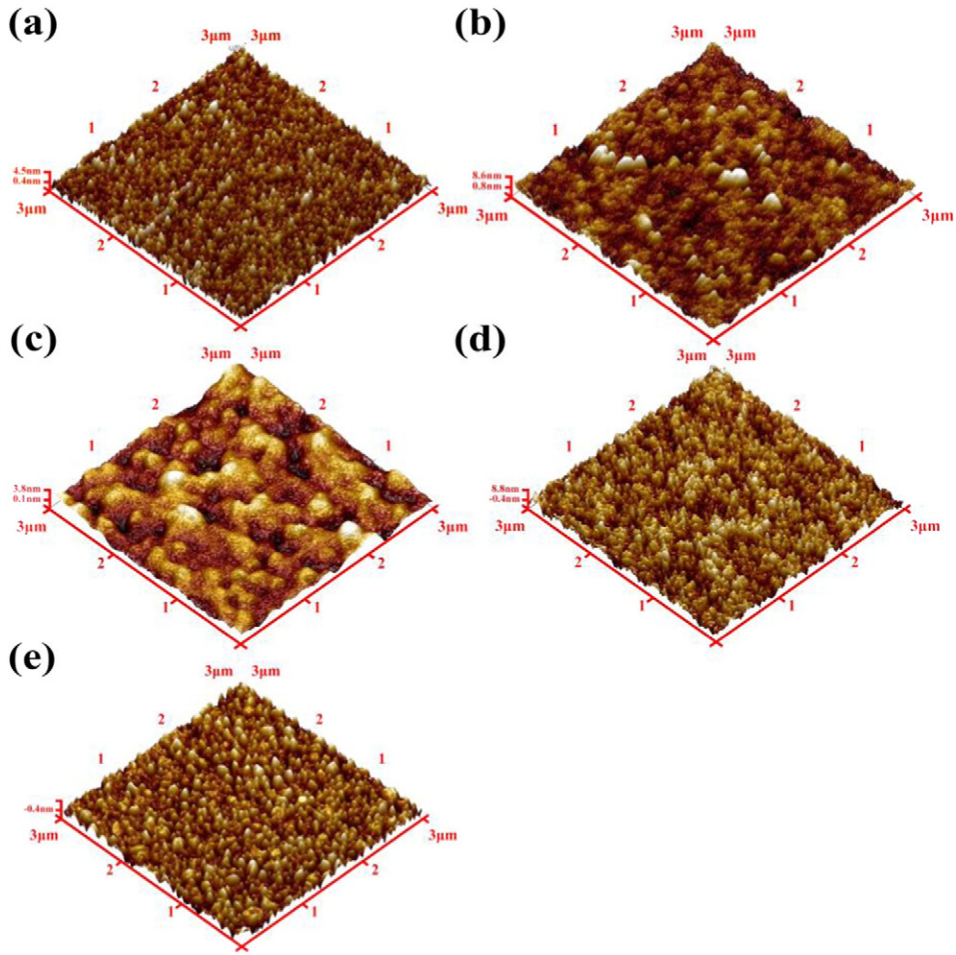


Fig. 2. 3D AFM images of (a) as-deposited  $Cs_{0.32}WO_3$  thin films and those annealed at (b) 300 °C, (c) 350 °C, (d) 400 °C, and (e) 450 °C.

The EDX spectra of the  $Cs_xWO_3$  thin films are shown in Fig. 4. Fig. 4(a) shows peaks corresponding to W and O, with no Cs on the surface, because of local area. Fig. 4(b) shows peaks corresponding to W, Cs, and O, indicating the possible formation of  $Cs_xWO_3$  on the surface.

Table 1 displays the element analysis results of the  $Cs_xWO_3$  before and after  $H_2$  annealing at different temperature by XPS. It can be found that as-deposited  $Cs_xWO_3$  thin film exhibits the minimum Cs/W

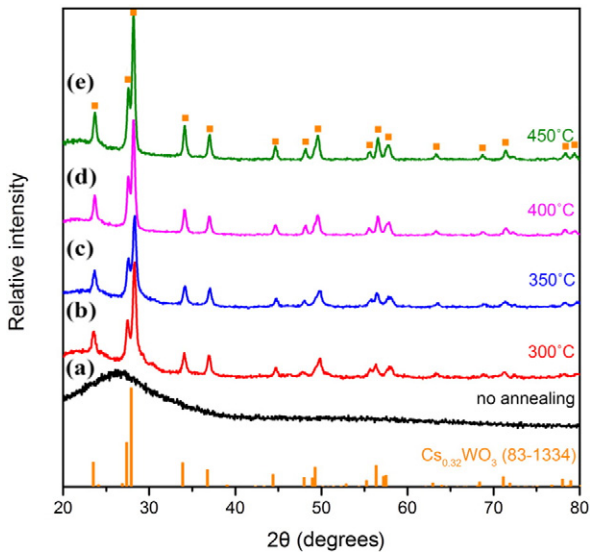


Fig. 3. XRD patterns of  $Cs_xWO_3$  thin films deposited on quartz glass substrates via electron beam evaporation: (a) as-deposited  $Cs_{0.32}WO_3$  thin films and annealed at (b) 300 °C, (c) 350 °C, (d) 400 °C, and (e) 450 °C.

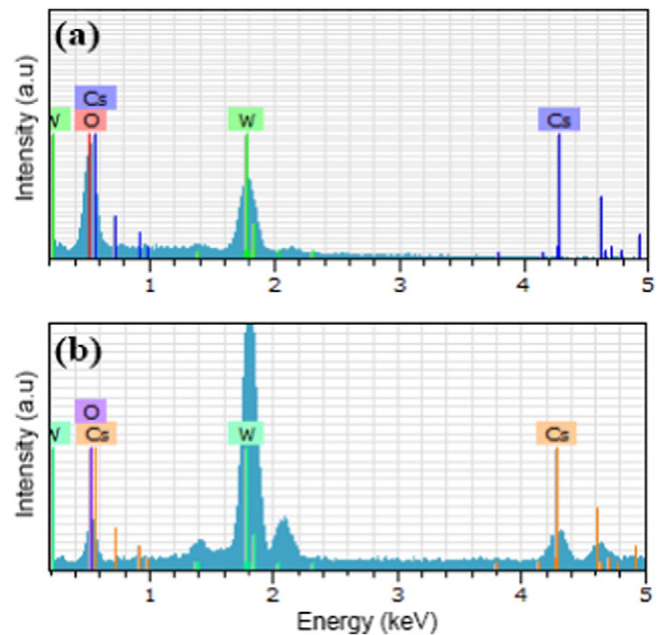


Fig. 4. EDX spectra of (a) as-sputtered film and (b) films annealed at above 300 °C.



**Table 1**  
The XPS analysis of  $\text{Cs}_x\text{WO}_3$  before and after  $\text{H}_2$  annealing at different temperatures.

| Sample                    | Element concentration (%) |                 |                  |                 | O/W atomic ratio | Cs/W atomic ratio | Composition                        |
|---------------------------|---------------------------|-----------------|------------------|-----------------|------------------|-------------------|------------------------------------|
|                           | C <sub>1s</sub>           | O <sub>1s</sub> | Cs <sub>3d</sub> | W <sub>4f</sub> |                  |                   |                                    |
| No $\text{H}_2$ annealing | 50.6                      | 33.0            | 4.3              | 12.1            | 2.73             | 0.36              | $\text{Cs}_{0.36}\text{WO}_{2.73}$ |
| $\text{H}_2$ -300°C       | 29.6                      | 47.3            | 7.1              | 16.0            | 2.96             | 0.44              | $\text{Cs}_{0.44}\text{WO}_{2.96}$ |
| $\text{H}_2$ -350°C       | 31.5                      | 44.8            | 8.6              | 15.2            | 2.95             | 0.57              | $\text{Cs}_{0.57}\text{WO}_{2.95}$ |
| $\text{H}_2$ -400°C       | 30.1                      | 46.8            | 9.0              | 14.1            | 3.32             | 0.64              | $\text{Cs}_{0.64}\text{WO}_{3.32}$ |
| $\text{H}_2$ -450°C       | 25.4                      | 47.5            | 11.2             | 15.9            | 2.99             | 0.70              | $\text{Cs}_{0.7}\text{WO}_{2.99}$  |

atomic ratio, which indicate the thin films existence of Cs element. Furthermore, the thin films after annealed at 300–450 °C exhibit higher Cs/W atomic ratio. This result is entirely consistent with previous studies [16].

Fig. 5(a) shows the W 4f doublet of the  $\text{Cs}_x\text{WO}_3$  thin films before and after  $\text{H}_2$  annealing, as obtained from XPS. The peaks of the  $\text{W}_{4f}$  core-level spectra of the as-sputtered film are located at 35.1 and 37.2 eV [22]. The two peaks of  $\text{W}_{4f}$  show a shift to higher binding energies after  $\text{H}_2$  annealing, which indicates that the chemical state of W changed due to the Cs-deficient surface of the  $\text{Cs}_x\text{WO}_3$  thin films after the reduction reaction. However, excess oxygen deficiencies remained on the surface.  $\text{Cs}^+$  ions diffused from the inside of the  $\text{Cs}_{0.32}\text{WO}_3$  thin

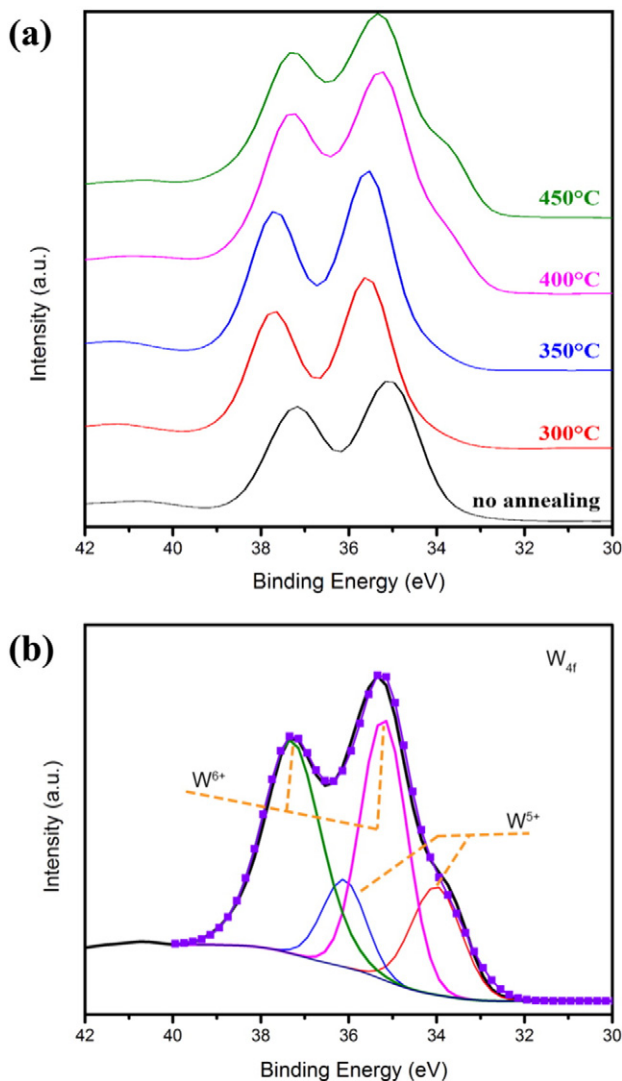
films to the surface during  $\text{H}_2$  annealing [23]. Fig. 5(b) shows the  $\text{W}_{4f}$  XPS spectra of the thin films after  $\text{H}_2$  annealing at 450 °C for 1 h. The  $\text{W}_{4f}$  core-level spectra can be fitted as two spin-orbit doublets,  $\text{W}_{4f_{7/2}}$  and  $\text{W}_{4f_{5/2}}$ , for the interval of about 2.1 eV. The peaks of the first doublet, located at 35.2 and 37.3 eV, are attributed to  $\text{W}^{6+}$  and those of the second doublet, located at 34 and 36.1 eV, are attributed to  $\text{W}^{5+}$ . The chemical state of W changed due to the partial reduction of  $\text{W}^{6+}$ . This suggests that the non-stoichiometric cesium tungsten bronzes  $\text{Cs}_x\text{WO}_3$  are reduced compounds, which can be expressed by the general formula  $\text{Cs}_x\text{W}_x^{5+}\text{W}_1^{6+} - x\text{O}_3$ .

### 3.3. Optical properties of films

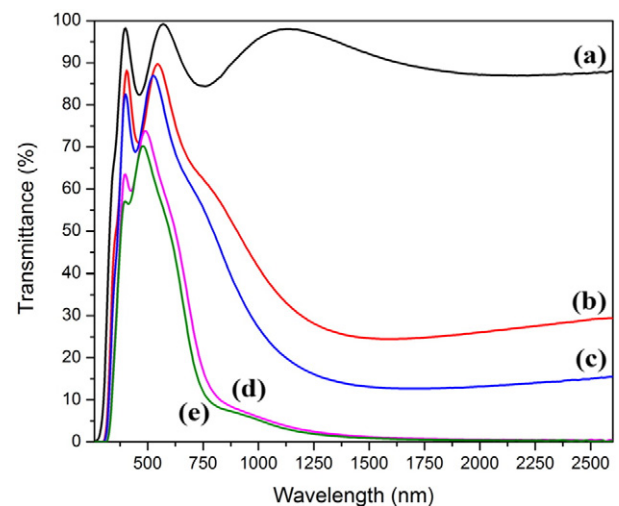
Fig. 6 shows the transmittance spectra of  $\text{Cs}_{0.32}\text{WO}_3$  thin films on quartz glass substrates before and after annealing at various temperatures. Fig. 6(a) shows that the  $\text{Cs}_{0.32}\text{WO}_3$  thin films before annealing are highly transparent in both the visible and NIR regions due to the wide band gap and lack of free electrons [24]. Figs. 6(b) and (c) show that the films are transparent pale blue and have more than 80% transmittance in the visible region. The films have high NIR light shielding performance as well as high transparency for visible light. When the annealing temperature was 400 °C, the transmittance was about 70% and also decrease in the NIR region, respectively. This indicates that the  $\text{Cs}_{0.32}\text{WO}_3$  thin films after annealing at 400–450 °C have better NIR shielding properties than those of films annealed at 300–350 °C. The sample annealed at 450 °C showed excellent NIR shielding ability as well as high transparency in the visible light region. The NIR shielding ability of the films is higher than that of ITO glass. This result indicates that  $\text{Cs}_{0.32}\text{WO}_3$  thin film is a promising heat shielding material that can replace ITO and ATO. A previous study [25] reported that  $\text{Cs}_{0.32}\text{WO}_3$  thin films realized stronger absorption properties of the NIR light rather than reflection. It has been suggested that NIR absorption is related to the plasmon resonance of free electron interband transition and small polarons [26,27]. The reduction of the film transmittance is due to the oxygen deficiency occurred at high annealing temperatures.

## 4. Conclusion

$\text{Cs}_x\text{WO}_3$  thin films were prepared via electron beam evaporation and annealed in an  $\text{H}_2$  atmosphere. The effects of annealing on the microstructure and NIR shielding properties of the  $\text{Cs}_x\text{WO}_3$  thin films were investigated. The films grown at RT were amorphous. Crystalline films were obtained at annealing temperatures of above 300 °C. The film color changed from pale yellow to deep blue with increasing annealing



**Fig. 5.** XPS spectra of  $\text{Cs}_x\text{WO}_3$  films prepared via electron beam evaporation: (a) as-sputtered and annealed at various temperatures and (b) 450 °C for 1 h.



**Fig. 6.** Transmission spectra of  $\text{Cs}_{0.32}\text{WO}_3$  thin films: (a) as-sputtered and annealed at (b) 300 °C, (c) 350 °C, (d) 400 °C, and (e) 450 °C.

temperature. The grain size and surface roughness of the films increased with increasing annealing temperature. The NIR shielding properties of  $\text{Cs}_x\text{WO}_3$  films were improved by annealing at 300–450 °C under a pure  $\text{H}_2$  atmosphere, with the amorphous thin films transforming into crystalline films. With annealing at 450 °C,  $\text{Cs}_x\text{WO}_3$  films showed the best NIR shielding properties, with a high transmittance of visible light (70%) and a high NIR shielding ratio (99%).

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